

## Spin transfer in Prussian blue, measured using polarised neutron diffraction

A.S. Wills<sup>1</sup>, E. Lelièvre-Berna<sup>2</sup>, F. Tasset<sup>2</sup>, J. Schweitzer<sup>3</sup>, R. Ballou<sup>4</sup>, A. Ludi<sup>6</sup>, P. Day<sup>2</sup>

<sup>1</sup>Department of Chemistry, University College London, 20 Gordon Street, London, WC1H 0AJ, UK

<sup>2</sup>Davy Faraday Research Laboratory, The Royal Institution of Great Britain, 21 Albemarle Street, London, W1S 4BS, UK

<sup>3</sup>Institut Laue-Langevin, BP 156, 38042 Grenoble Cedex 9, France

<sup>4</sup>Département de Recherche Fondamentale sur la Matière Condensée, SPSMS, CEA Grenoble, 38054 Grenoble, France

<sup>5</sup>Laboratoire de Magnetisme Louis Néel - CNRS, BP 166, 38042 Grenoble Cedex 9, France

<sup>6</sup>Departement für Chemie und Biochemie, Universität Bern, Freiestrasse 3, 3000 Bern

Prussian Blue,  $\text{Fe}^{\text{III}}_4[\text{Fe}^{\text{II}}(\text{CN})_6]_3 \cdot x\text{H}_2\text{O}$ , is the archetypal mixed-valent molecular ferromagnet. It displays ferromagnetic ordering of the high spin Fe(III) spins below  $T_c=5.5$  K. Arguments based on the charge transfer transition in the optical absorption spectrum at  $14100\text{ cm}^{-1}$  that gives rise to its well known intense blue colour, led Day and coworkers to search for spin transfer from the high-spin Fe(III) to the nominally diamagnetic Fe(II) using polarised neutron diffraction.[1] Technical limitations at the time restricted their measurements to only 3 reflections. 30 years on, using state-of-the-art nuclear polarised  $^3\text{He}$  technology, we re-examined the same sample of deuterated Prussian blue and demonstrate conclusively the existence and sign of spin transfer from  $\text{Fe}^{3+}$  to the  $\text{Fe}^{2+}$ .

### References

[1] - P. Day, F. Herren, A. Ludi *et al.*, *Helvetica Chimica Acta* **63**, 148 (1980).

[2] A.S. Wills, E. Lelièvre-Berna, F. Tasset, J. Schweizer, R. Ballou, *Physica B* **356**, 254 (2004).